

Passive materials selection (CRAs) versus corrosion inhibitor development for CO₂ transport and injection

DCR6

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Background and Research Questions

CCS infrastructure requires reliable CO₂ transport and injection in harsh environments.

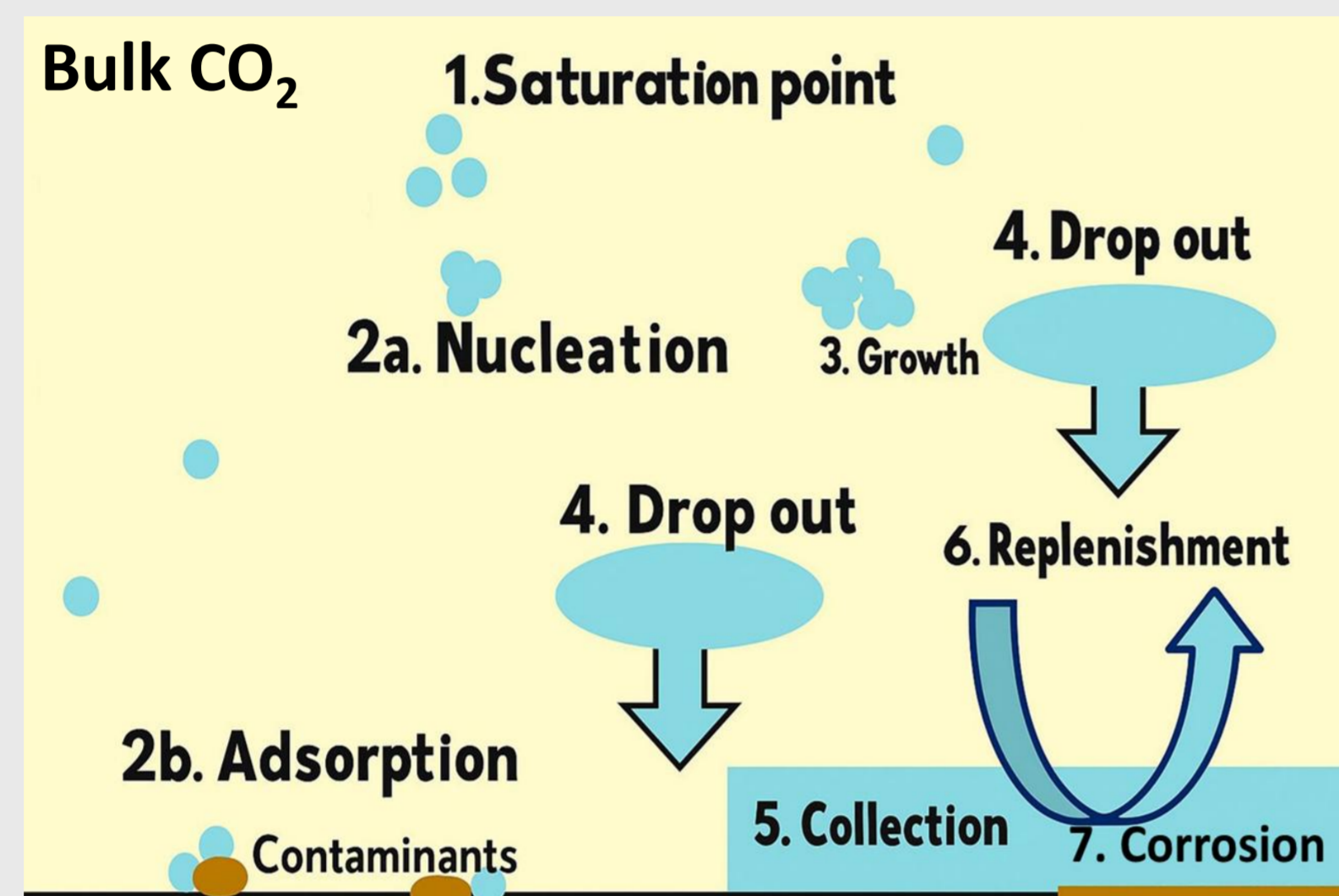
- What are the most effective **corrosion mitigation strategies** in **acidic, wet, and high-pressure environments**? → **Passive materials (CRAs) and corrosion inhibitors**
- How do **CO₂ stream impurities** interact with pipeline materials and influence corrosion?
- How do **corrosion inhibitors and CRAs** work under such conditions?

Secondments



Transport pipelines (X65)

- Analysis of different levels of **water content and impurities** in the form of **NO_x, SO_x, and O₂**;
- **Acid drop-out**, [1].



Representation of the different steps of corrosion due to formation and drop-out of liquids in a bulk CO₂ phase, [1].

Injection pipeline (CRAs) Alleima

- High-salinity brine flowback from a saline aquifer;
- Super 13Cr;
- SAF 2507, super-duplex stainless steel;
- SANICRO 28, high-alloy multi-purpose austenitic stainless steel;
- SANICRO 35, alloy combining the best features of a super austenitic stainless steel and a nickel alloy.

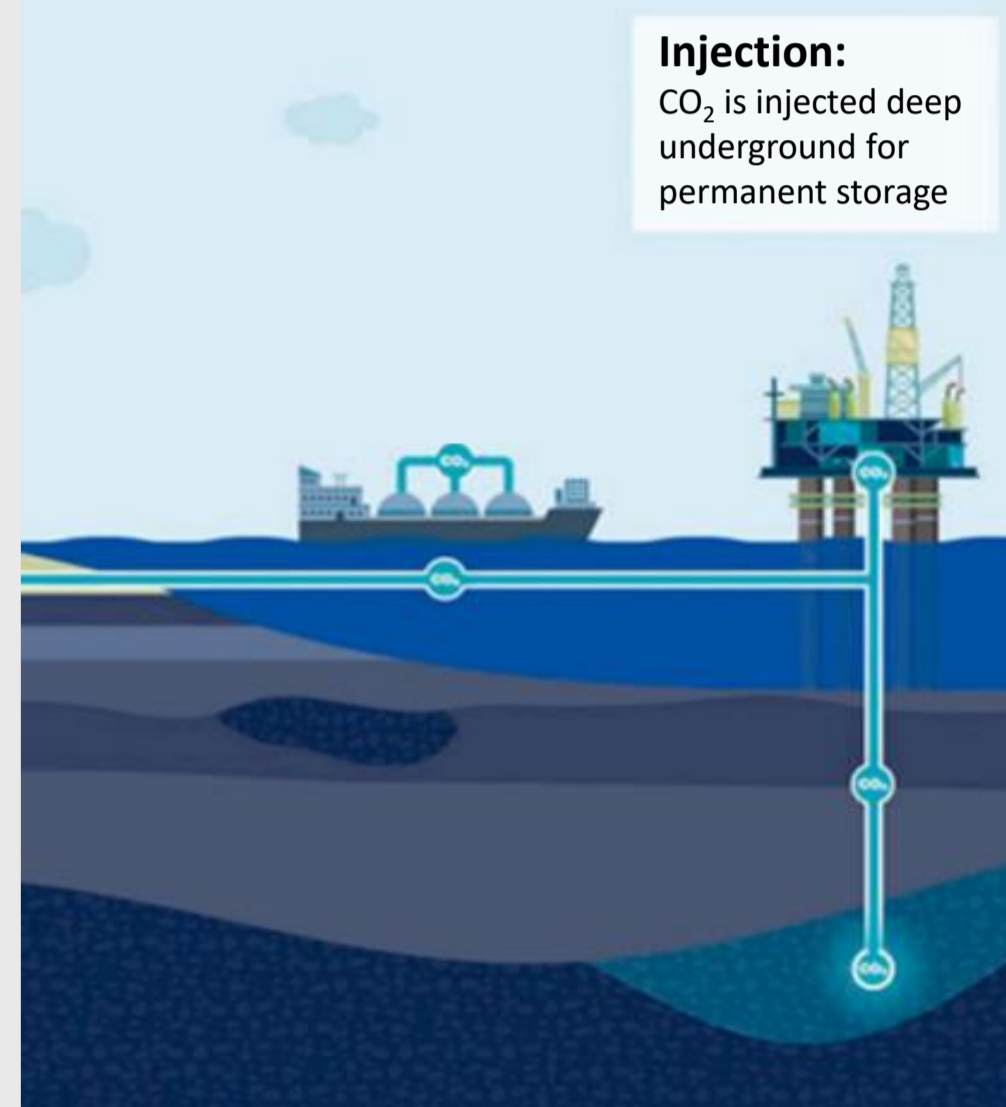


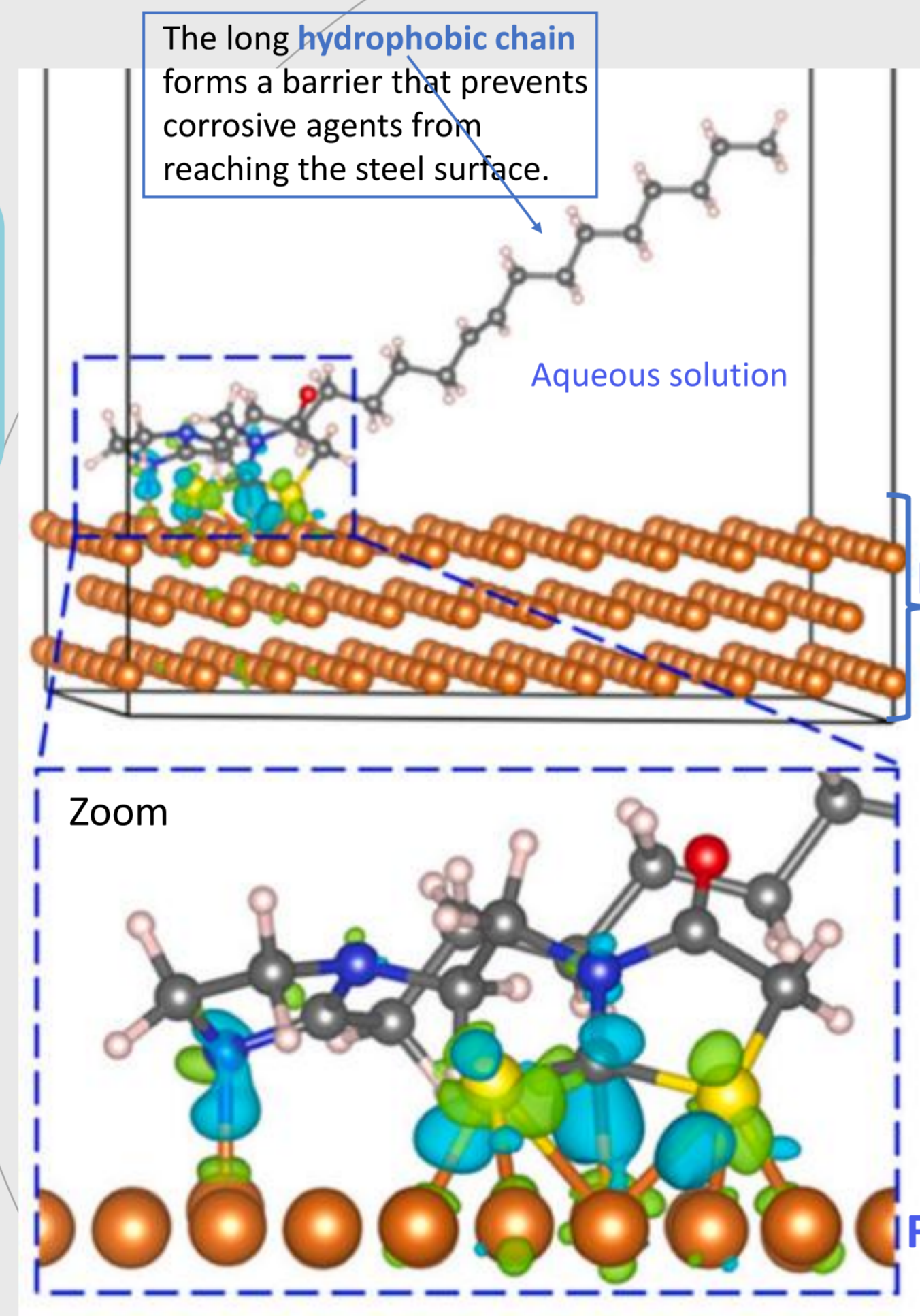
Image from S.K. Kairy, S. Zhou, A. Turnbull, G. Hinds, Corrosion of pipeline steel in dense phase CO₂ containing impurities: A critical review of test methodologies, Corrosion Science (2023) 214 110986.

Inhibitors selection



- **Chemical structure** → Role of heteroatoms in the imidazole ring;
- **Solubility** → Sapphire cell; DTU
- **Interaction with impurities** → H₂O, O₂, SO_x, NO_x
- **Degradation products**;
- **Adsorption mechanism** → Efficiency; (Optimal concentration)
- **Film's stability** → Inhibitors blend.

*Some inhibitors provided by Baker Hughes, others from literature reviews.



Adsorption of imidazoline derivative corrosion inhibitors on Fe surface through N and S atoms, [2].

CRA passive film stability

- Influence of impurities on the **passive plateau**, [3];
- Properties of **passive film**;
- **Chloride content** → Breakdown of passivity;
- **pH** → Depassivation;
- **Ability to passivate** → Repassivation mechanism;
- **Pitting resistance** → PREN;
- **Crevice corrosion** → Design.

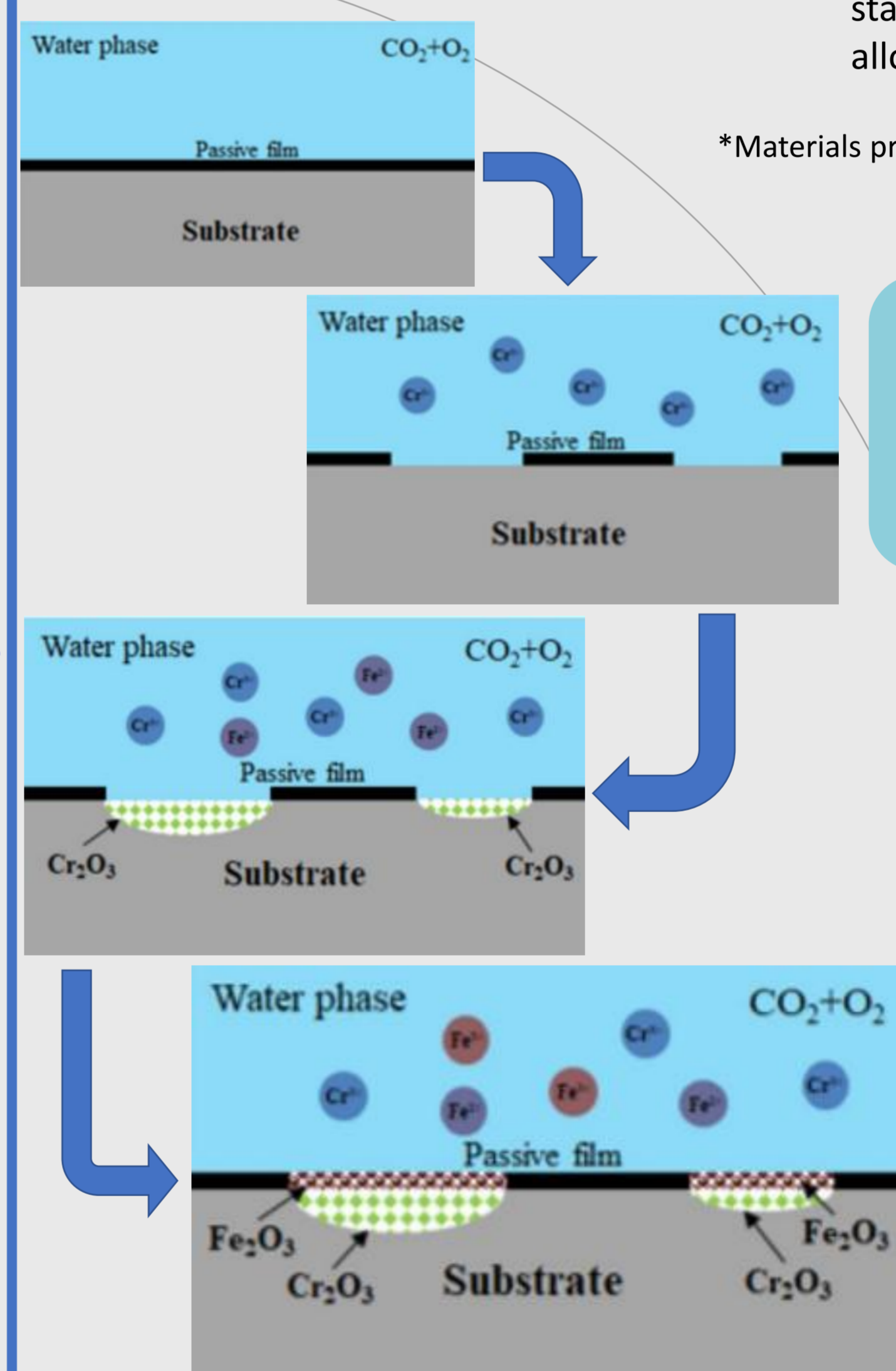


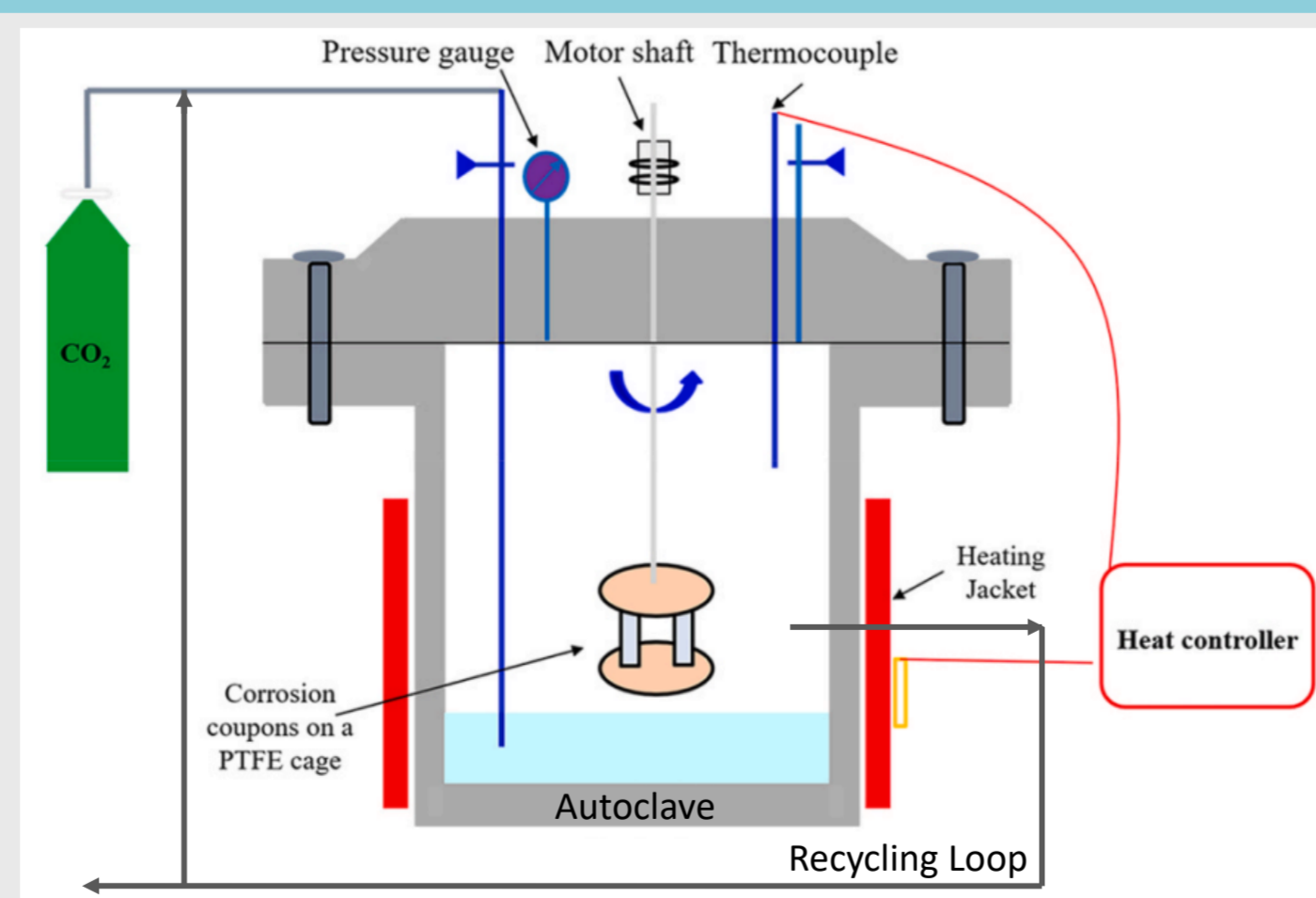
Diagram of the depassivation mechanism of Super 13Cr at high temperature, high CO₂ content, [5].

Experimental Methodology: Autoclave Conditions and Aqueous Solutions

- **P > 80 bar**;
- **pH measurement** with a **ZrO₂-based** high temperature high pressure pH electrode.

First Step → **Static** autoclave;

Second Step → **Dynamic** Conditions: **Rotation Mechanism and Recycling Loop**.



Supercritical CO₂ corrosion experimental device diagram.

- **P atm**;
- Use of **Aqueous Solutions** to investigate corrosion mechanism using **classical electrochemical techniques**. → **Birbilis' Protocol**, [4].

Standard three-electrodes set-up

The **working electrode** is a **rotating disk electrode (RDE)** with an exposed disk surface area of ~1 cm², made of the testing material (steel or CRAs).

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