

MECHANISTIC INSIGHTS INTO LONG-TERM INTEGRITY OF CEMENT AND STEEL-CEMENT INTERFACE IN CO₂ STORAGE FACILITIES

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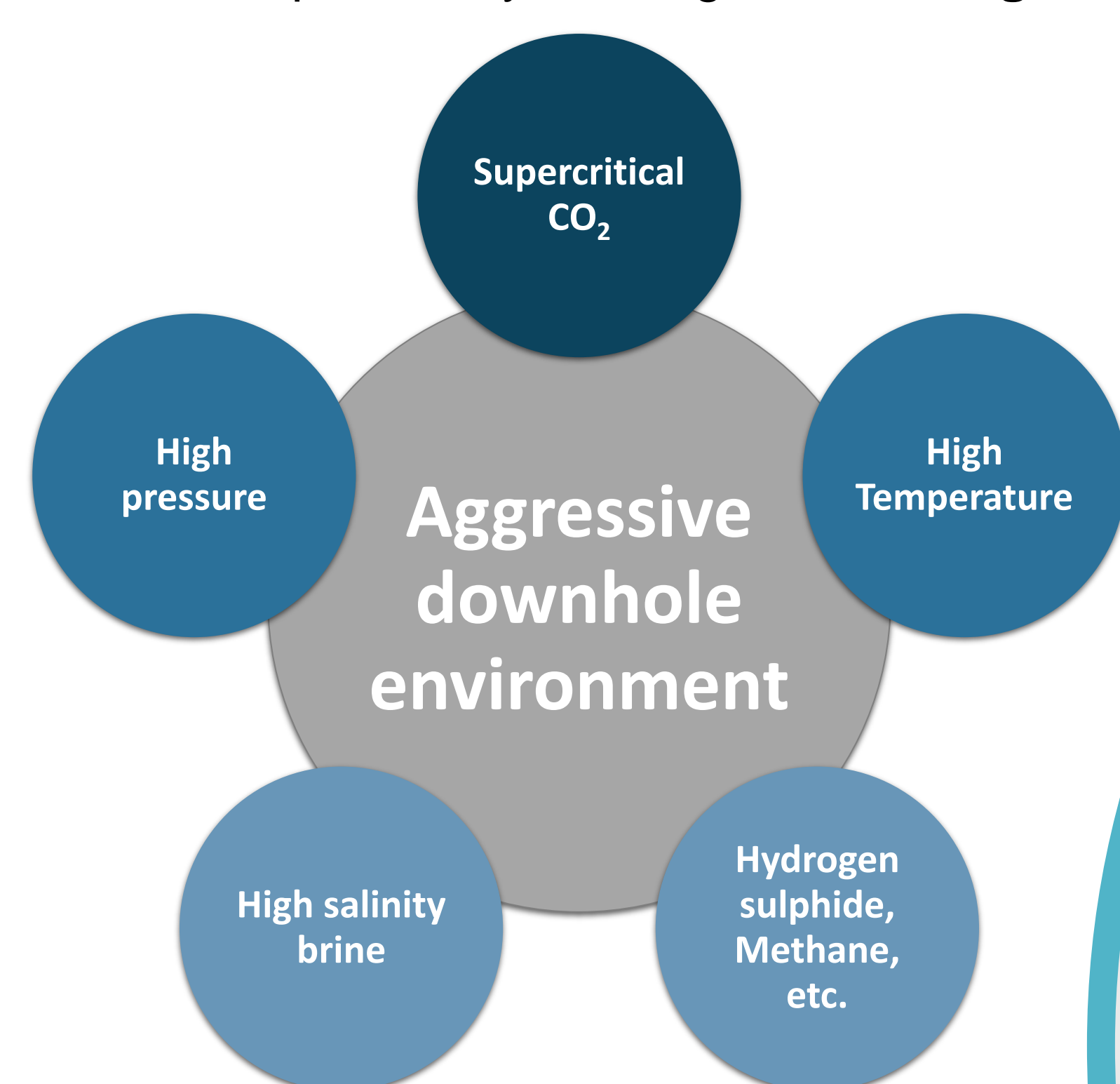
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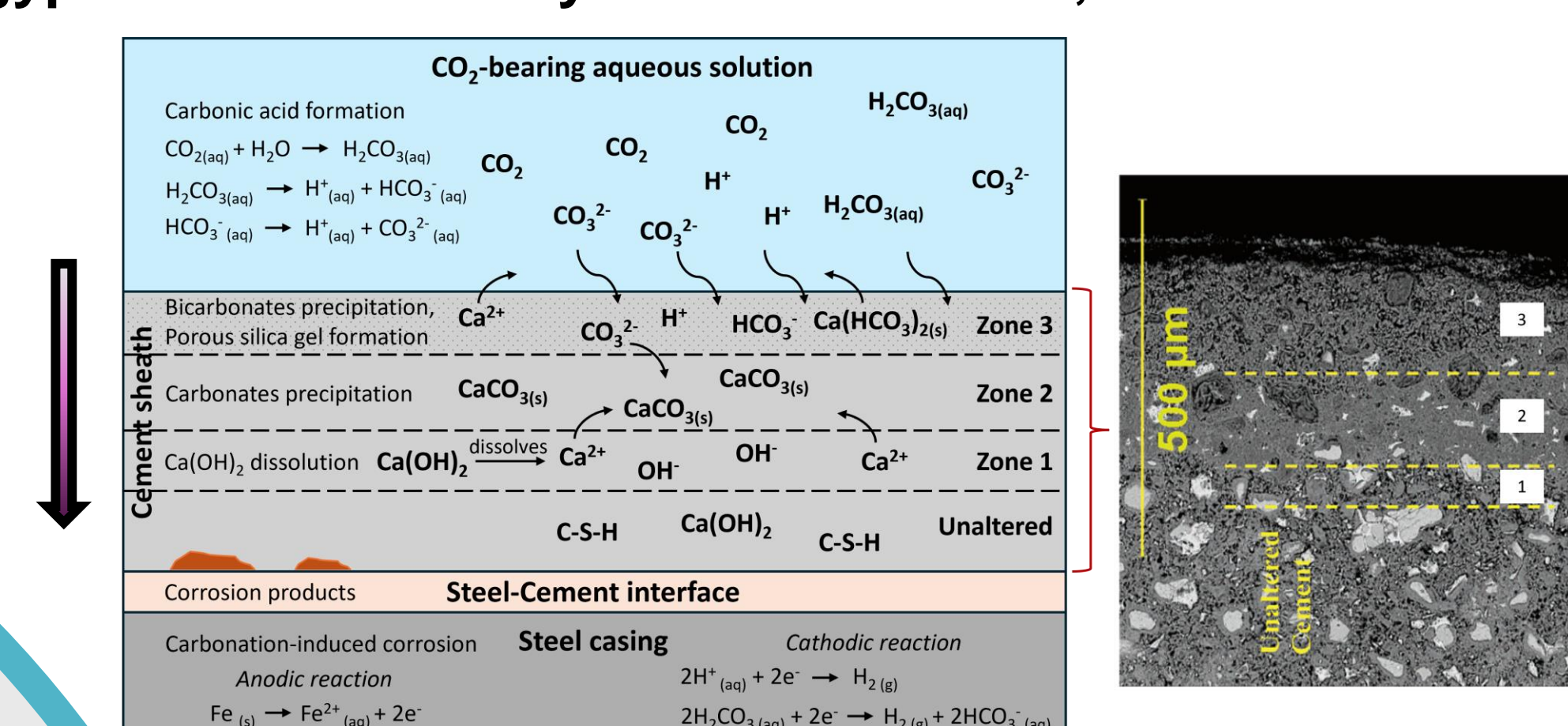
Background

- **Long-term wellbore integrity** with low risk of carbon dioxide (CO₂) leakage is the key to safe and effective CO₂ storage
- **CO₂ leakage through steel-cement (S-C) system** is reported to occur much more rapidly than geological leakage through formation rock in a CO₂ storage reservoir [1]
- Exposure of cement to sCO₂ or CO₂-bearing brines may lead to **carbonation of cement**, potentially causing **cement degradation** and **corrosion of steel**



Degradation Mechanism

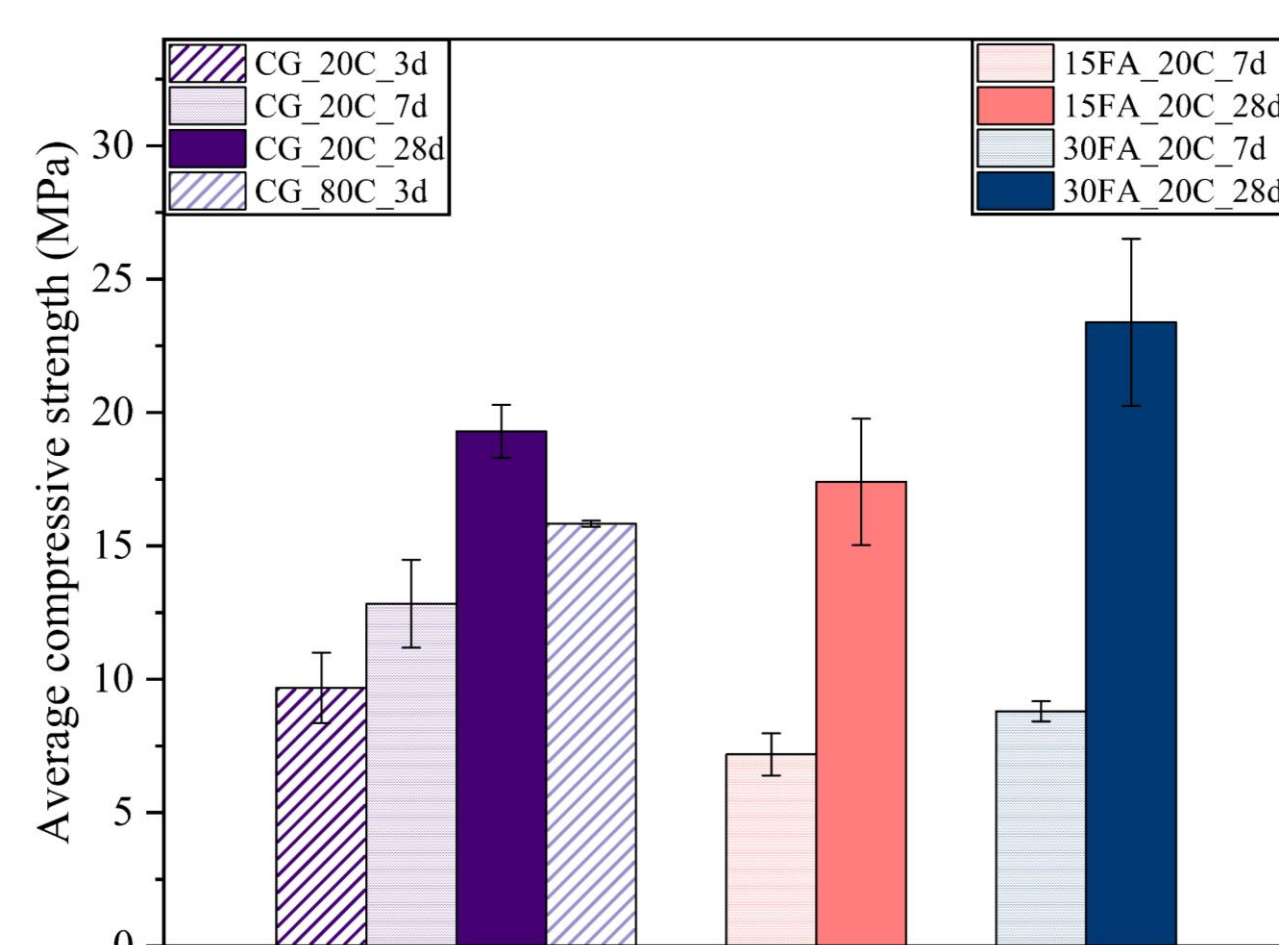
- Initially calcium carbonate precipitation may lead to **pore refinement** and improved mechanical strength
- **Over extended exposure**, due to calcium leaching and C-S-H dissolution, **porosity increases**, and **strength reduces**
- Thermodynamic end state of carbonated cement slurry is an assemblage of **calcium carbonate, gypsum and oxide/ hydroxides of silica, alumina and iron**



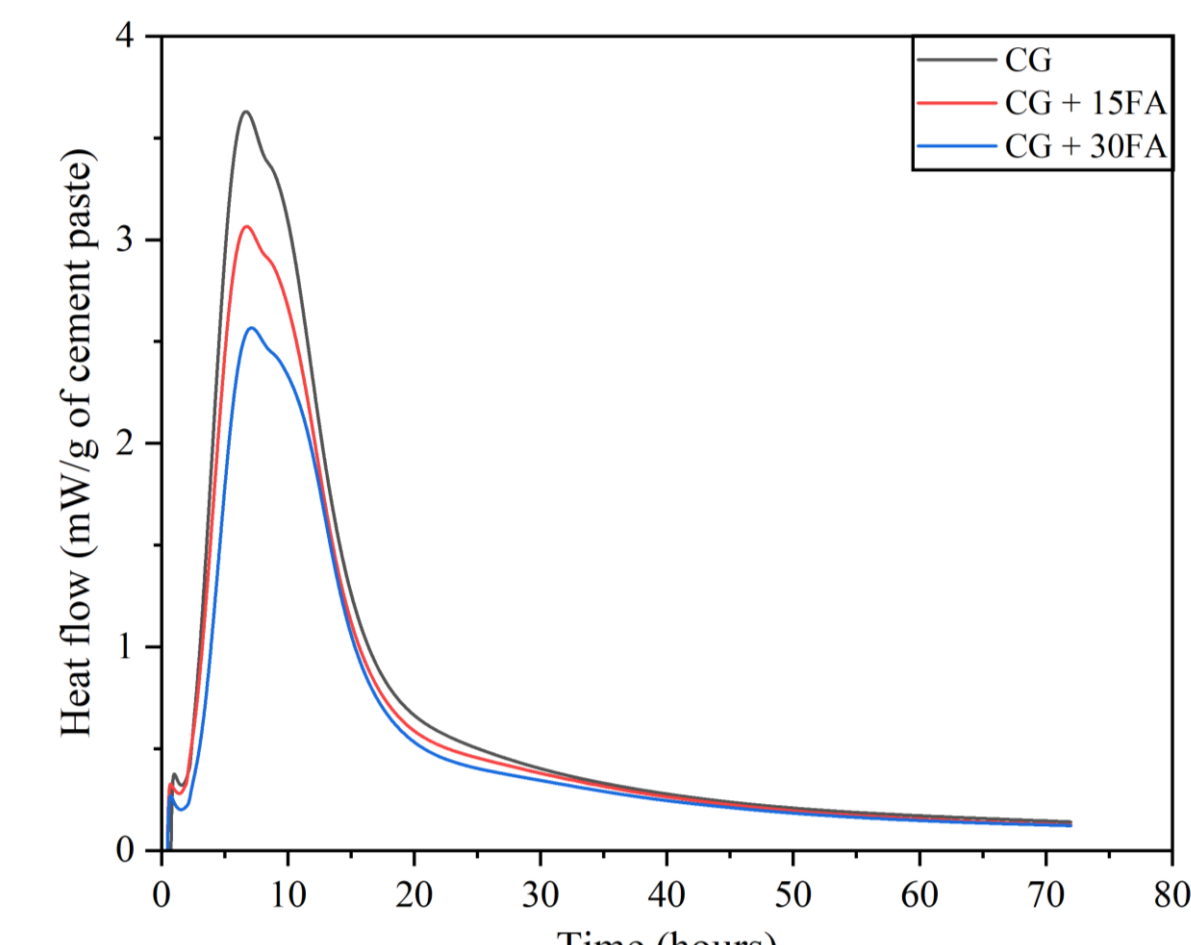
Research Gaps

- Limited understanding of the effect of **high temperature (HT) & high pressure (HP) curing for longer duration** on hydration and carbonation kinetics of well cements
- Contradictory findings on the **use of composite cements** to improve long-term durability [3,4,5]. No clear agreement on the practical dosage of mineral additives
- No clarity on the effect of carbonation-induced changes on **permeability and strength of cement**
- Lack of **representative test methodology** to characterise durability performance and its relationship to long-term durability

Preliminary Results



Average compressive strength of cement samples cured at 20 °C and 80 °C for different ages



Heat flow curves of cement samples tested at 38 °C for 3 days in an Isothermal Calorimeter

Aim & Objectives

Effect of **water-saturated sCO₂ & sCO₂-saturated brine** on integrity of cement and S-C interface at simulated wellbore conditions

1 Critical Curing conditions

- Degree of hydration and hydration kinetics
- Phase assemblage
- Mechanical strength

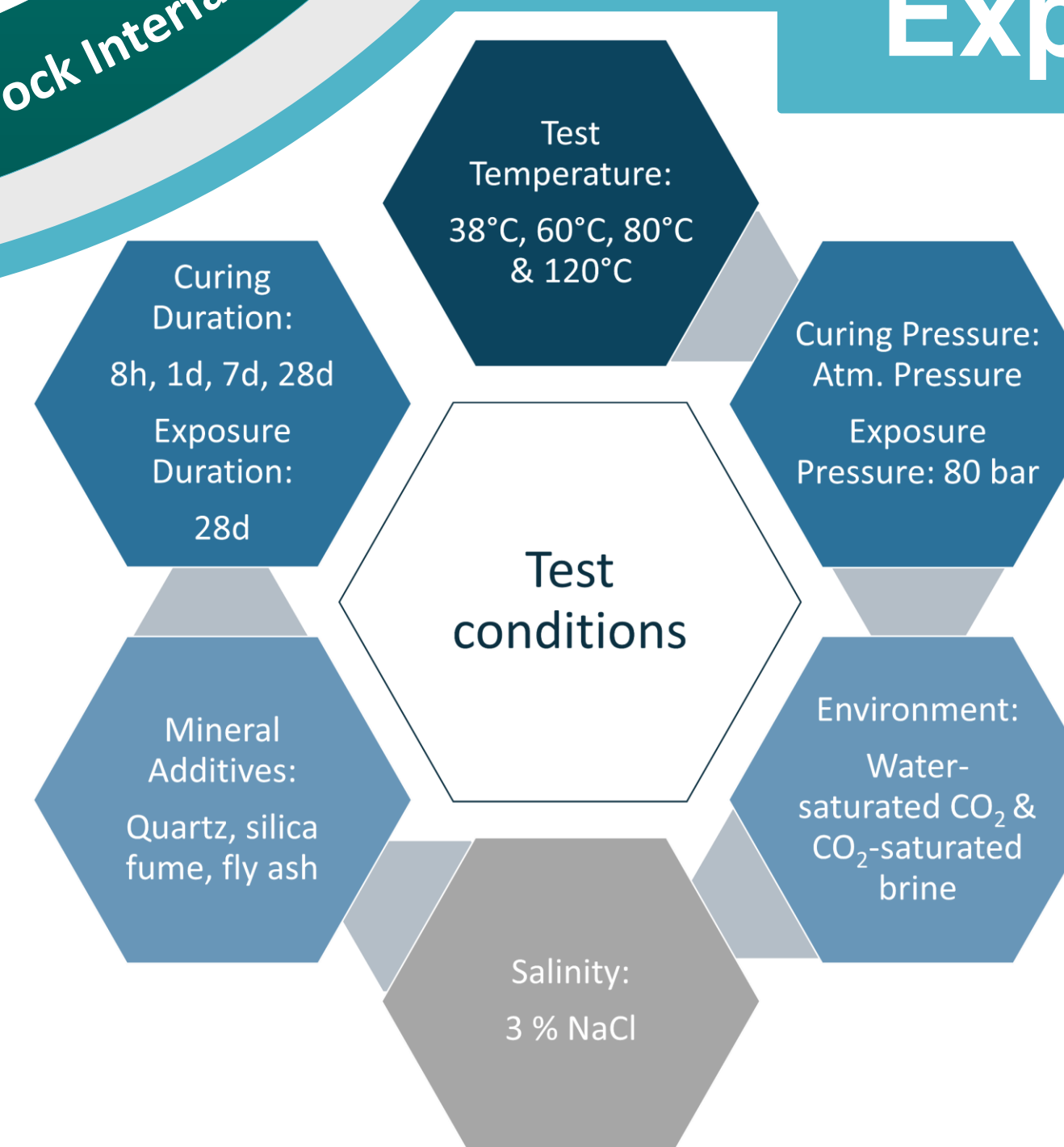
2 Cement-Environment interaction

- Physico-chemical changes
- Mechanical strength
- Durability properties

3 Integrity of S-C interface

- Corrosion performance in simulated conditions
- Physical and chemical characteristics at the S-C interface
- Bond strength at interface after 6 months of steel corrosion

Experimental Plan



Autoclave for tests in simulated well conditions

Hydration & phase assemblage

- Evolution of hydration products at different curing ages (XRD, TGA)
- Hydration kinetics at different HT (Isothermal calorimetry)
- Carbonation reaction products

Mechanical properties

- Compressive strength at different curing ages and after carbonation
- Strength retrogression at HT (120 °C)

Transport & durability properties

- Porosity – before & after carbonation (SEM)
- Water Sorptivity
- Bulk resistivity
- Carbonation depth

Corrosion performance

- Electrochemical sensing technology using autoclaves
- Electrochemical Impedance Spectroscopy (EIS)
- Linear Polarisation resistance (LPR)

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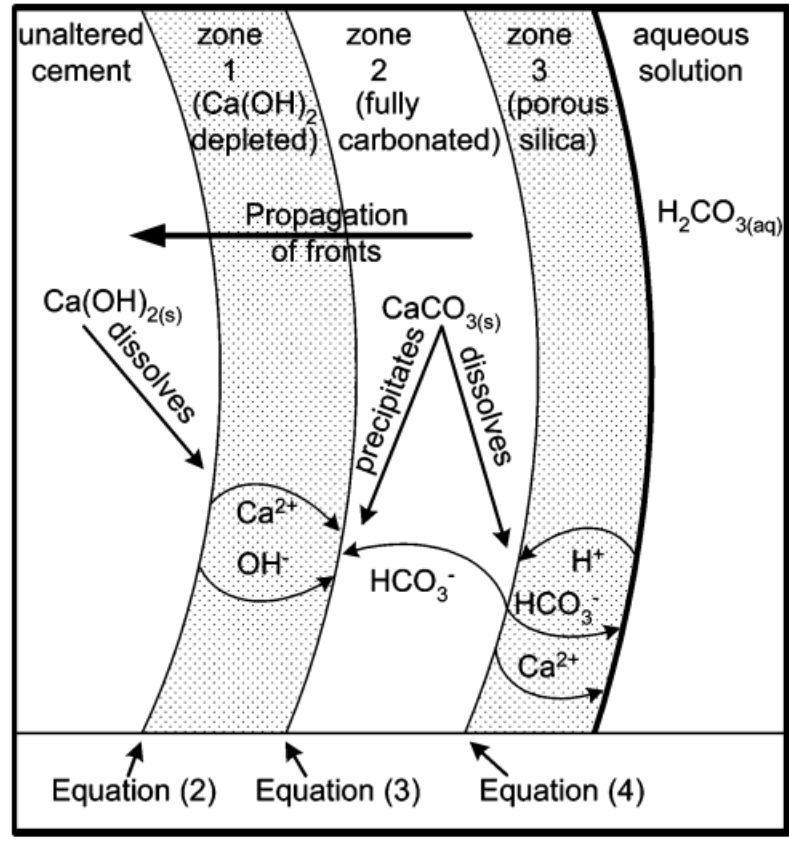
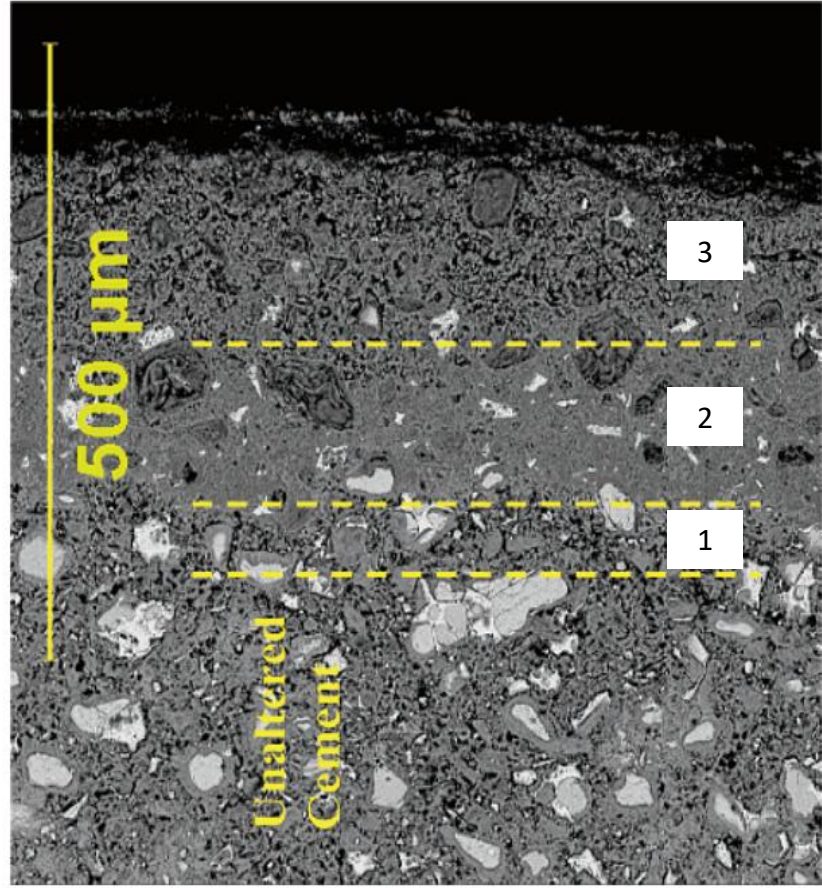
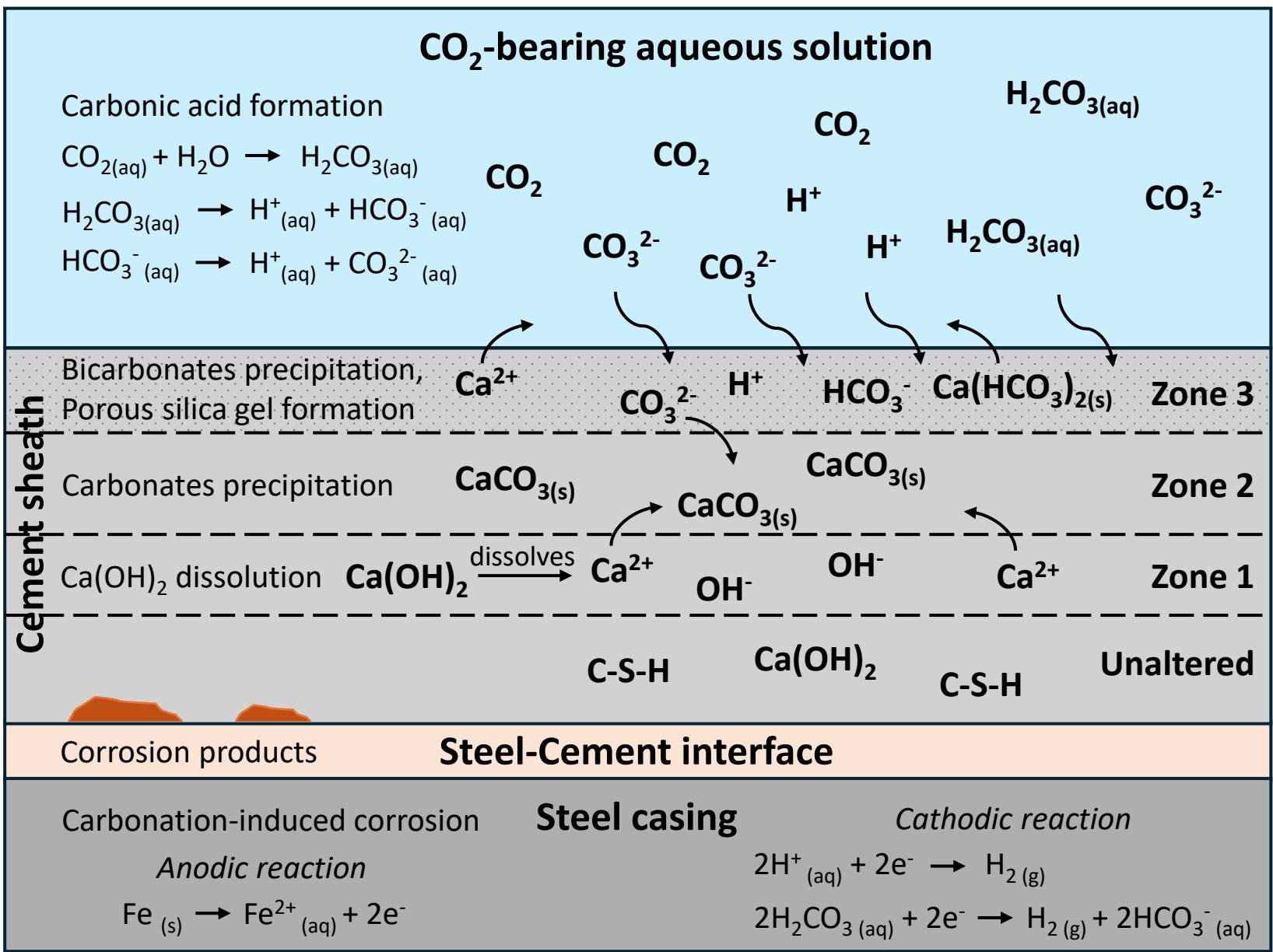


FIGURE 3. Schematic showing the dissolution and calcium migration and formation of distinct zones in the cement. Equation numbers refer to chemical reactions that occur at each front as detailed in the text. Note: in the cement structure, local pH is buffered by $\text{CaCO}_{3(s)}$ and $\text{Ca(OH)}_{2(s)}$.



1

Critical Curing conditions

- Degree of hydration and hydration kinetics
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